

Page 01 of 06



# Zinc Chloride, anhydrous

# **SECTION 1. IDENTIFICATION**

Product Identifier Zinc Chloride, anhydrous

Other Means of

Zinc Dichloride

Identification

Product Code(s) ZI1210

Product Family Inorganic Solid

Recommended Use Laboratory and industrial use.

Restrictions on Use None known.

Supplier Identifier Alphachem Limited, 2485 Milltower Court, Mississauga, Ontario, L5N 5Z6, (905) 821-2995

Emergency Phone No. CANUTEC CANADA, 613-996-6666, 24 Hours

SDS No. 0244

# **SECTION 2. HAZARD IDENTIFICATION**

Classified according to Canada's Hazardous Products Regulations (WHMIS 2015) and the US Hazard Communication Standard (HCS 2012).

#### Classification

Acute toxicity (Oral) - Category 4; Skin corrosion - Category 1; Serious eye damage - Category 1; Specific target organ toxicity (single exposure) - Category 3

# **Label Elements**





Signal Word: Danger

Hazard Statement(s):

Causes severe skin burns and eye damage.

May cause respiratory irritation.

Harmful if swallowed.

Precautionary Statement(s):

Wear protective gloves/protective clothing/eye protection/face protection.

Do not get in eyes, on skin, or on clothing.

Wash hands thoroughly after handling.

Use only outdoors or in a well-ventilated area.

Avoid breathing dust/fume/gas/mist/vapours/spray.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF ON SKIN: Wash with plenty of water.

If skin irritation or rash occurs: Get medical advice/attention.

IF INHALED: Remove person to fresh air and keep comfortable for breathing.

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

Continue rinsing.

Immediately call a POISON CENTRE or doctor.

In case of fire: Evacuate area.

Fight fire with normal precautions from a reasonable distance.

Storage:

Store in a well-ventilated place. Keep container tightly closed.

# Other Hazards

None known.

# SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS

### Substance:

Chemical Name	CAS No.	%	Other Identifiers
Zinc chloride	7646-85-7	>94	Zinc Dichloride

# **SECTION 4. FIRST-AID MEASURES**

### First-aid Measures

### Inhalation

Remove source of exposure or move to fresh air. Keep at rest in a position comfortable for breathing. DO NOT move about unnecessarily. Symptoms of pulmonary edema may be delayed.

#### Skin Contact

Avoid direct contact. Wear chemical protective clothing if necessary. Immediately rinse skin with lukewarm, gently flowing water for at least 30 minutes. DO NOT INTERRUPT FLUSHING. If it can be done safely, continue flushing during transport to hospital.

### Eye Contact

Avoid direct contact. Wear chemical protective gloves if necessary. Immediately rinse the contaminated eye(s) with lukewarm, gently flowing water for at least 30 minutes, while holding the eyelid(s) open. Take care not to rinse contaminated water into the unaffected eye or onto the face. Immediately call a Poison Centre or doctor. Specific treatment is required.

### Ingestion

Never give anything by mouth if person is rapidly losing consciousness, or is unconscious or convulsing. Do not induce vomiting. Rinse mouth with water.

### **First-aid Comments**

Some of the first-aid procedures recommended here require advanced first-aid training. If exposed or concerned, get medical advice or attention.

# Most Important Symptoms and Effects, Acute and Delayed

If on skin: may burn the skin. Permanent scarring may result. If in eyes: may cause serious eye damage. May irritate or burn the eyes. Permanent damage including blindness may result.

# **Immediate Medical Attention and Special Treatment**

# Special Instructions

General advice

Consult a physician. Show this safety data sheet to the doctor in attendance. Move out of dangerous area.

# **SECTION 5. FIRE-FIGHTING MEASURES**

# Extinguishing Media

### Suitable Extinguishing Media

Not combustible. Use extinguishing agent suitable for surrounding fire.

# Unsuitable Extinguishing Media

None known.

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015 Page 02 of 06

### Specific Hazards Arising from the Product

Closed containers may rupture violently when heated releasing contents. Heating increases the release of toxic vapour.

In a fire, the following hazardous materials may be generated: corrosive hydrogen chloride; toxic zinc oxides.

# Special Protective Equipment and Precautions for Fire-fighters

Evacuate area. Fight fire from a safe distance or a protected location. Approach fire from upwind to avoid hazardous vapours or gases. For a massive fire, immediately evacuate the area and use unmanned hose holder or monitor nozzles.

Fire-fighters should enter area wearing specialized protective equipment. (Bunker Gear will not provide adequate protection.) chemical protective clothing (e.g. chemical splash suit) and positive pressure SCBA may be necessary.

# SECTION 6. ACCIDENTAL RELEASE MEASURES

# Personal Precautions, Protective Equipment, and Emergency Procedures

Evacuate the area immediately. Isolate the hazard area. Keep out unnecessary and unprotected personnel. Do not touch damaged containers or spilled product unless wearing appropriate protective equipment. Increase ventilation to area or move leaking container to a well-ventilated and secure area. Eliminate all ignition sources if safe to do so. Remove or isolate incompatible materials as well as other hazardous materials.

### **Environmental Precautions**

Do not allow into any sewer, on the ground or into any waterway. If the spill is inside a building, prevent product from entering drains, ventilation systems and confined areas.

# Methods and Materials for Containment and Cleaning Up

Stop or reduce leak if safe to do so. Contain and soak up spill with absorbent that does not react with spilled product. Place used absorbent into suitable, covered, labelled containers for disposal. Collect using shovel/scoop or approved HEPA vacuum and place in a suitable container for disposal. Flush spill area. Large spills or leaks: contact emergency services and manufacturer/supplier for advice.

# **SECTION 7. HANDLING AND STORAGE**

# Precautions for Safe Handling

Wear personal protective equipment to avoid direct contact with this chemical. Avoid heating that will increase the amount of vapours. Prevent accidental contact with incompatible chemicals. Avoid generating vapours or mists. Only use where there is adequate ventilation. Keep containers tightly closed when not in use or empty.

# Conditions for Safe Storage

Store in an area that is: cool, dry, out of direct sunlight and away from heat and ignition sources, separate from incompatible materials (see Section 10: Stability and Reactivity). Restrict access to authorized personnel only. Keep amount in storage to a minimum. Empty containers may contain hazardous residue. Store separately. Keep closed. Follow all precautions given on this safety data sheet. Comply with all applicable health and safety regulations, fire and building codes.

# SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

### Control Parameters

Chemical Name	ACGIH TLV®		OSHA PEL		AIHA WEEL	
	TWA	STEL	TWA	Ceiling	8-hr TWA	TWA
Zinc chloride	Not established		Not established			

### **Appropriate Engineering Controls**

Use a corrosion-resistant exhaust ventilation system separate from other ventilation systems. Exhaust directly to the outside, taking any necessary precautions for environmental protection. Use local exhaust ventilation and enclosure, if necessary, to control amount in the air.

### **Individual Protection Measures**

# Eye/Face Protection

Wear chemical safety goggles and face shield when contact is possible.

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015 Page 03 of 06

#### Skin Protection

Wear chemical protective clothing e.g. gloves, aprons, boots.

No specific guidelines are available. Contact chemical manufacturer/supplier for advice.

### Respiratory Protection

Wear a NIOSH approved air-purifying respirator with N100, R100, or P100 filter(s), wear a NIOSH approved air-purifying respirator with an appropriate cartridge.

# SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

# **Basic Physical and Chemical Properties**

Appearance White crystalline powder.

Odour Odourless
Odour Threshold Not available
pH 4.0 (10% solution)

Melting Point/Freezing Point 290 °C (554 °F) (melting); 290 °C (554 °F) (freezing)

Initial Boiling Point/Range 732 °C (1350 °F)
Flash Point Not applicable
Evaporation Rate Not applicable
Flammability (solid, gas) Will not burn.

Upper/Lower Flammability or

**Explosive Limit** 

Not applicable (upper); Not applicable (lower)

Vapour Pressure
Not available
Vapour Density (air = 1)
Relative Density (water = 1)
Not applicable
2.907 at 25 °C

**Solubility** Very soluble (more than 50 g/100 mL) in water; Soluble in all proportions in

alcohols (e.g. ethanol).

Partition Coefficient,

n-Octanol/Water (Log Kow)

0.15 (estimated)

Auto-ignition Temperature Not applicable

Decomposition Temperature Not available

Viscosity Not applicable (kinematic); Not applicable (dynamic)

Other Information

Physical State Solid

### SECTION 10. STABILITY AND REACTIVITY

### Reactivity

Not reactive under normal conditions of use.

Chemical Stability
Normally stable.

Possibility of Hazardous Reactions

None known.

Conditions to Avoid

High temperatures.

Incompatible Materials

Strong bases (e.g. sodium hydroxide).

**Hazardous Decomposition Products** 

None known.

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015 Page 04 of 06

# SECTION 11. TOXICOLOGICAL INFORMATION

# Likely Routes of Exposure

Inhalation; skin contact; eye contact; ingestion.

# **Acute Toxicity**

Chemical Name	LC50	LD50 (oral)	LD50 (dermal)
Zinc chloride		> 1000 mg/kg (male rat)	

### Skin Corrosion/Irritation

Contact can cause pain, redness, burns, and blistering. Permanent scarring can result.

### Serious Eye Damage/Irritation

Causes serious eye damage based on skin corrosion information.

# STOT (Specific Target Organ Toxicity) - Single Exposure

### Inhalation

May cause severe nose and throat irritation, severe lung injury.

# Skin Absorption

No information was located.

### Ingestion

Toxic, can cause death Causes damage to organs severe irritation or burns to the mouth, throat and stomach. Symptoms may include nausea, vomiting, stomach cramps and diarrhea.

# **Aspiration Hazard**

No information was located.

# STOT (Specific Target Organ Toxicity) - Repeated Exposure

Causes dermatitis. Symptoms may include dry, red, cracked skin (dermatitis).

# Respiratory and/or Skin Sensitization

No information was located for respiratory sensitization. Not a skin sensitizer.

### Carcinogenicity

Chemical Name	IARC	ACGIH®	NTP	OSHA
Zinc chloride	Not evaluated	Not designated	Not Listed	

### Reproductive Toxicity

### **Development of Offspring**

No information was located.

# Sexual Function and Fertility

No information was located.

### Effects on or via Lactation

No information was located.

# **Germ Cell Mutagenicity**

No information was located.

# Interactive Effects

No information was located.

# **SECTION 12. ECOLOGICAL INFORMATION**

This section is not required by WHMIS. This section is not required by OSHA HCS 2012.

# **SECTION 13. DISPOSAL CONSIDERATIONS**

### **Disposal Methods**

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015 Page 05 of 06

Dispose of contents and container in accordance with local, regional, national and international regulations.

# SECTION 14. TRANSPORT INFORMATION

Regulation	UN No.	Proper Shipping Name	Transport Hazard Class(es)	Packing Group
Canadian TDG	UN2331	Zinc chloride, anhydrous	8	III
US DOT	UN2331	Zinc chloride, anhydrous	8	Ш

Special Precautions Not applicable

Transport in Bulk According to Annex II of MARPOL 73/78 and the IBC Code

Not applicable

### SECTION 15. REGULATORY INFORMATION

Safety, Health and Environmental Regulations

Canada

Domestic Substances List (DSL) / Non-Domestic Substances List (NDSL)

Listed on the DSL.

USA

Toxic Substances Control Act (TSCA) Section 8(b)

Listed on the TSCA Inventory.

# **SECTION 16. OTHER INFORMATION**

SDS Prepared By Alphachem Limited
Phone No. (905)-821-2995

Date of Preparation November 26, 2015

Date of Last Revision April 29, 2016

References CHEMINFO database. Canadian Centre for Occupational Health and Safety (CCOHS).

Disclaimer This document is offered only as a guide in the safe handling of the above product, and have a product of the above product.

This document is offered only as a guide in the safe handling of the above product, and has been prepared from the best information currently available. It is not intended to be all-inclusive and the conditions of use may involve other additional considerations. Since Alphachem Limited cannot anticipate or control the conditions under which the product may be used, it will not be liable for any claims, damages or losses which may result from the use or

reliance on any information herein.

Product Identifier: Zinc Chloride, anhydrous

Date of Preparation: November 26, 2015 Page 06 of 06